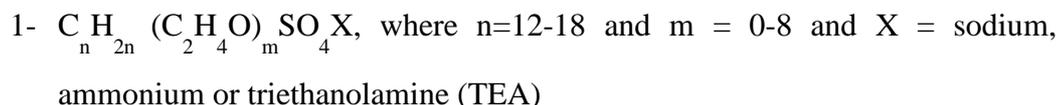


## Sodium laureth sulphate

Sodium laureth sulphate or sodium lauryl ether sulfate (SLES) is an example of anionic surfactant. It belongs to the Alcohol ethoxysulphates (AES) group. AES are a widely used class of anionic surfactants and found in household cleaning and personal care products. The structure of AES could be described by any of the following formulas:



As part of the PBT assessment of sodium laureth sulphate data from structurally analogous substances (other AES) will be used to fulfil some endpoints for the assessed substance (SDA, 1991; HERA, 2003).

An assessment of the PBT status of Sodium laureth sulphate using the algorithm described in the toolbox is explained below step by step.

### Step 1: Substance identification

Identifiers on sodium laureth sulphate have been obtained from the most recent reliable sources ([www.chemfinder.com](http://www.chemfinder.com)) which is listed in [step1](#). The smiles notation for sodium laureth sulphate was obtained from [chemspider](#) and chemID plus source. These identifiers are shown in Table1 below.

Identifier of Sodium laureth sulphate	
EINECS or ELINCS number	239-925-1
CAS name and CAS number	9004-82-4
Name(s) in the IUPAC nomenclature or other international chemical name(s)	Sodium dodecylpoly(oxyethylene) sulphate
Other names (usual name, trade name, abbreviation)	Sodium laureth sulfate; Dodecyl alcohol, monoether with polyethylene glycol, hydrogen sulfate sodium salt; Sodium polyethoxyethyl dodecylsulfate; Polyethylene glycol, mono(hydrogen sulfate), dodecyl ether, sodium salt; Poly(oxy-1,2-ethanediyl), alpha-sulfo-omega-(dodecyloxy)-, sodium salt; Poly(oxyethylene) lauryl ether sulfate



the [www.heraproject.com](http://www.heraproject.com). Some data related to SLES were reported in this report since it belongs to ASE category.

3. The data gap which is highlighted as red in the table.

Required Property	Value	Source
Water solubility	Very soluble 187mg/l (estimate)	HSDB and HERA (2003)
Partition coefficient n-octanol/water	No measured logK <sub>ow</sub> is available. Log K <sub>ow</sub> = 1.62 (calculated)	HERA (2003)
Soil Adsorption Coefficient (Koc/Kd)	No measured log Koc value is available for Sodium laureth sulphate.	
Henry's Law Constant	No experimentally determined Henry's law constant information is available.	

**Table2:** Physical chemical data available for Sodium laureth sulphate.

## 2- Accumulation data of Sodium laureth sulphate

No accumulation data were provided for Sodium laureth sulphate. However, an environmental assessment report has been published by the Danish EPA (Masden et al, 2001), which contains studies carried out to determine the BCF values for analogue alkyl ether sulfates. Based on these studies, the report concludes that AES is not considered to bioconcentrate in aquatic organisms.

## 3- Degradation data of Sodium laureth sulphate

The following table provides a summary of the

1. Degradation properties required for the assessment as explained in the algorithm.
2. The available degradation data of Sodium laureth sulphate along with the source for these data. As explained in the algorithm, the data should be reliable.
3. The data gap which is highlighted as red in the table.

Required Property	Value	Source
Biodegradation	On the basis of the available data, Sodium laureth sulphate is considered to be readily biodegradable under aerobic and anaerobic conditions and easily degradable both primarily and ultimately  (study conducted according to OECD TG 301C guideline)	HERA (2003)
<b>Half lives-t<sub>1/2</sub></b>		
1-Hydrolysis as a function of pH  2-Photolysis (Atmospheric OH Rate Constant)	No experimental aquatic degradation data are available for Sodium laureth sulphate.  No experimental data are available for Sodium laureth sulphate only calculated one Rate Cons = 1.67 *10 <sup>-11</sup>	

**Table3:** Degradation data available for Sodium laureth sulphate.

#### 4- Environmental Partitioning (MacKay)

No data was available.

#### 5- Aquatic toxicity information of Sodium laureth sulphate

Aquatic toxicity data for structurally similar compounds have been identified in several sources (HERA risk assessment, Masden et al, 2001). The data are summarised below in Table 4.

Required Property	Value	Source
<b>Fish</b> Acute toxicity to fish (96hrs LC <sub>50</sub> ) mg/l	The 24-96 h LC <sub>50</sub> values for seven different fish species range between 0.39 mg/L to 450 mg/L. The toxicity of AES with chain length <C <sub>16</sub> decreases with increasing numbers of EO groups and peaks at chain lengths of C <sub>16</sub> .	HERA, Masden2001
Long term toxicity to fish (28days NOEC) mg/l		
<b>Daphnia</b> Acute toxicity to Daphnia (48hrs EC <sub>50</sub> ) mg/l	For Daphnia magna, the acute EC <sub>50</sub> values range between 1 and 50 mg/L were reported for AES.	
Long term toxicity to Daphnia (21days NOEC) mg/l		
<b>Algae</b> Acute toxicity to algae (72hrs EC <sub>50</sub> ) mg/l	For algae, typical EC <sub>50</sub> values range from 4 to 65 mg/L. For AES with chain lengths in the same range as Sodium laureth sulphate, the 21 day EC <sub>50</sub> is 20 mg/L for C <sub>12-14</sub> AE <sub>n</sub> S, the 72 h EC <sub>50</sub> is 32 mg/L for C <sub>12-14</sub> AES and the 48 h EC <sub>50</sub> is 65 mg/L for C <sub>10-15</sub> AE <sub>3</sub> S.	HERA, Masden2001

**Table4:** Aquatic toxicity data available for Sodium laureth sulphate.

#### Step 4: Filling the data gap by using QSAR

In this step the above endpoints (both the available and not) will be predicted using QSARs tools and software listed in step 4 of the algorithm (EPIWIN, Danish(Q)SAR data base and PBT profiler). The reason for doing this is to compare the QSAR results with the experimental one to identify the accuracy of the QSAR.

##### 1- Results obtained by using EPIWIN

Table 5 provides the predicted values for the above endpoints using EPIWIN software along with the name of the programme used. The output obtained by EPIWIN for each end point is given. Also included are explanation which in the toolbox would be obtained by clicking on a link to view. **An interesting point found in this step.** As explained in the algorithm that one way to enter the chemical into EPIWIN program is by typing its CAS registry number. An automatic look-up function based on the CAS Registry number is then automatically retrieving the chemical's SMILES notation using a pre-existing database containing over 100,000 records (SMILESCAS

database). However, in our case when entering the CAS NO of 9004-82-4 corresponding to Sodium laureth sulphate, a different SMILES notation appeared and therefore not the right chemical ([click here to view an example](#)). Accordingly, to get the correct estimate for Sodium laureth sulphate using EPIWIN the chemical was entered into the program by typing its SMILES notation; [Na+].[O-]S(=O)(=O)OCCOCCCCCCCCCCCCC.

Required Property	EPI QSAR Programme	Predicted Value
Water solubility	WSKOW <a href="#">(result output)</a>	486.5 mg/l at 25 C
Partition coefficient n-octanol/water	KOWWIN	Log K <sub>ow</sub> = 1.42
Bioconcentration Factor (BCF)	BCFWIN	LogBCF = 1.85 (BCF =70.79)
Soil Adsorption Coefficient (K <sub>oc</sub> /K <sub>d</sub> )	(PCKOCWIN)	K <sub>oc</sub> = 3489
Henry's Law Constant	HENRYWIN	2.86 *10 <sup>-9</sup> atm-m <sup>3</sup> /mole
<b>Half lives-t<sub>1/2</sub></b>		
1-Hydrolysis as a function of pH	HYDROWIN	Can not be estimated
2-Photolysis (Atmospheric OH Rate Constant)	AOPWIN	Atmospheric Oxidation Rate Constant is approximately around 31.46 E-12 cm <sup>3</sup> /molecule-sec Half-Life = 4hrs
( Atmospheric Oxidation, Ozone)	AOPWIN	Can not be estimated
<b>Biodegradability</b>		
BIOWIN1	BIOWIN	0.35 (Does not Biodegrade fast)
BIOWIN2	BIOWIN	0.36 (Does not biodegrade fast)
BIOWIN3 (Ultimate biodegradation)	BIOWIN	2.75 (weeks)
BIOWIN4 (Primary Biodegradation)	BIOWIN	3.63 (Days-weeks)
BIOWIN5	BIOWIN	0.37 (Not readily Degradable)
BIOWIN6	BIOWIN	0.17 (Not readily Degradable)
BIOWIN7	BIOWIN	0.52 ( biodegrades fast)
Ready Biodegradability Prediction:	BIOWIN	NO
<b>Environmental Partitioning</b>		

(MacKay)	EPI V3.2 (Results output)	Level III Fugacity Model: <table border="1"> <thead> <tr> <th></th> <th>Mass Amount (percent)</th> <th>Half-Life (hr)</th> <th>Emissions (kg/hr)</th> </tr> </thead> <tbody> <tr> <td>Air</td> <td>1</td> <td>8.6</td> <td>1000</td> </tr> <tr> <td>Water</td> <td>36</td> <td>360</td> <td>1000</td> </tr> <tr> <td>Soil</td> <td>64</td> <td>720</td> <td>1000</td> </tr> <tr> <td>Sediment</td> <td>0</td> <td>3.24*10<sup>3</sup></td> <td>0</td> </tr> </tbody> </table>		Mass Amount (percent)	Half-Life (hr)	Emissions (kg/hr)	Air	1	8.6	1000	Water	36	360	1000	Soil	64	720	1000	Sediment	0	3.24*10 <sup>3</sup>	0
	Mass Amount (percent)	Half-Life (hr)	Emissions (kg/hr)																			
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Water	36	360	1000																			
Soil	64	720	1000																			
Sediment	0	3.24*10 <sup>3</sup>	0																			
<b>Fish</b> Acute toxicity to fish (96hrs LC50) mol/l Long term toxicity (28days NOEC) mol/l	ECOSAR	Can not be estimated																				
<b>Daphnia</b> Acute toxicity to Daphnia (48hrs EC <sub>50</sub> ) mol/l Long term toxicity (21days NOEC) mol/l	ECOSAR	Can not be estimated																				
<b>Algae</b> Acute toxicity to Algae (72hrs EC <sub>50</sub> ) mol/l	ECOSAR	Can not be estimated																				

**Table5:** EPIWIN predictions of the required endpoints

## 2- Results obtained from Danish(Q)SAR database

Unfortunately, no data was found for Sodium laureth sulphate using the Danish(Q)SAR database. The search revealed no matches. However other QSAR databases could be used such as [chemspider](#) or [PBT profiler](#). Chemspider provides an online access to structure based predictions of systematic identifiers and physicochemical based predictions. You only need to type the name of your chemical, click on search and the predictions result will appear. The problem in using this source is that information about the QSARs models used for the prediction is not available. On the other hand, PBT Profiler is a no-cost computer based tool to screen chemicals lacking experimental data in order to help identify if they are potentially may persist, bioaccumulate, and be toxic to aquatic life, i.e., PBT chemicals. The good thing with PBT profiler, is that it provides a straight-forward estimate of persistence, bioaccumulation, and aquatic toxicity based on widely accepted criteria along with the explanation of the results. It is important to know that the PBT Profiler is a screening level predictive tool and cannot be used for all chemical substances.

Similarly to EPIWIN program finding, when entering Sodium laureth sulphate into PBT profiler via its CAS NO (9004-82-4), a different structure to Sodium laureth sulphate was found. The PBT profiler report for a chemical with CAN no of 9004-82-4 was for sodium lauryl ether sulfate as shown in figure1. In fact, this was expected

since PBT profiler is linked to the same SMILESCAS database that EPIWIN program make use of.

**Data Entry**

Estimate the persistence, bioaccumulation, and toxicity of SODIUM LAURYL ETHER SULFATE by starting the PBT Profiler

**Or**

Build the list of chemicals to be profiled by adding another CAS Registry number or other identifier:

[Draw your chemical](#)

---

**List of Chemicals to be Profiled**

#	CAS Number	Name	SMILES	
1	9004-82-4	SODIUM LAURYL ETHER SULFATE	[Na]OS(=O)(=O)OC=COC=COCCCCCCC CCCCC	<input type="button" value="X"/>

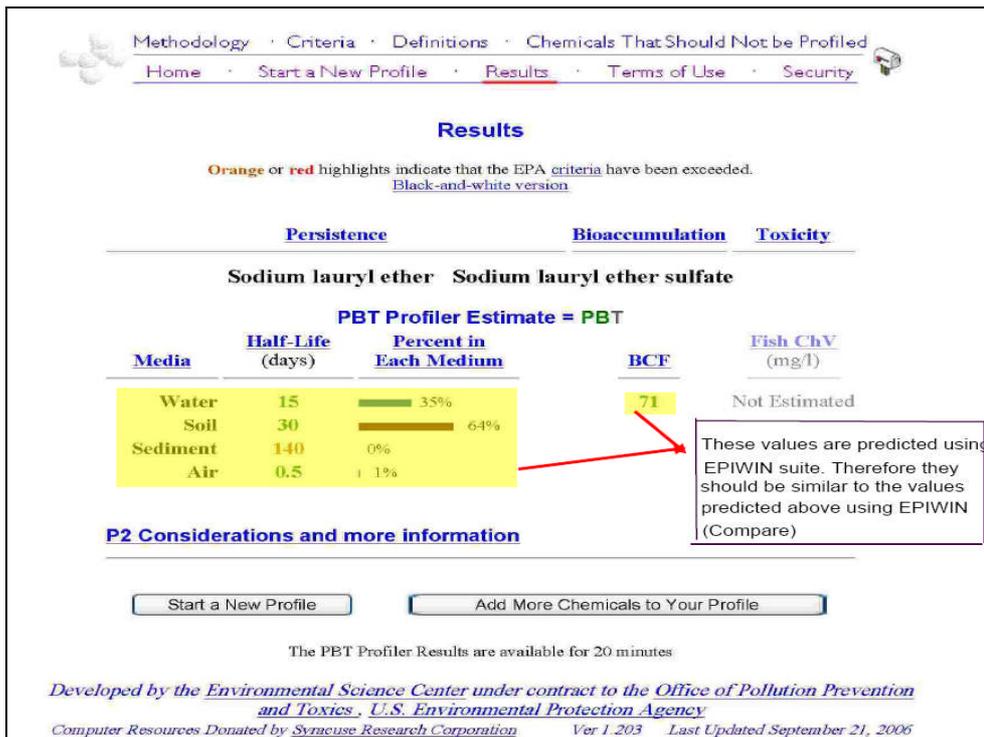
Not correct SMILES for chemical with CAS NO of 9004-82-4

[Black-and-white version](#)

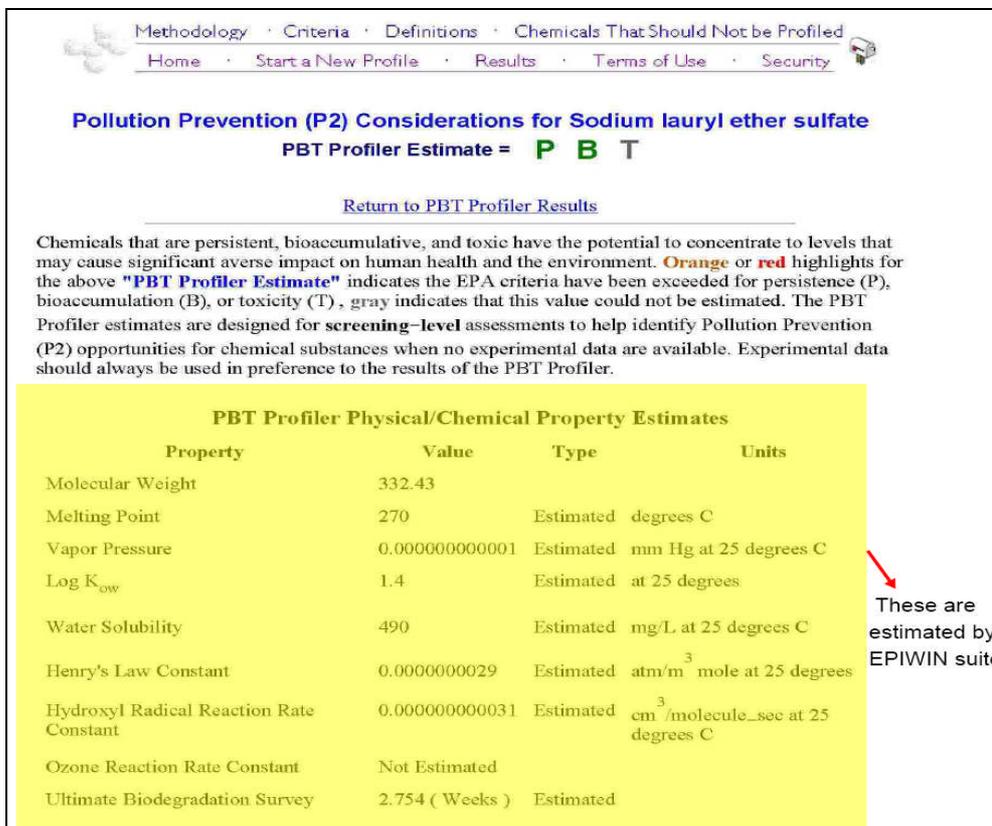
Developed by the *Environmental Science Center* under contract to the *Office of Pollution Prevention and Toxics*, U.S. Environmental Protection Agency  
Computer Resources Donated by *Syracuse Research Corporation* Ver 1.203 Last Updated September 21, 2006

**Figure 1:** PBT profiler data entry output for a chemical with CAS No of 9004-82-4.

To get the PBT profiler report for Sodium laureth sulphate, the chemical was entered into the tool via its name and SMILES notation. The PBT profiler report for Sodium laureth sulphate with some comments is shown in the Figures below. These outputs contain predictions for environmental Partitioning (MacKay), BCF, fish chronic toxicity (Figure2) along with some physical/chemical properties (Figure3).



**Figure2:** PBT profiler report output for Sodium laureth sulphate.



**Figure3:** PBT profiler report output for Sodium laureth sulphate.

Before proceeding to step 5, it is useful to compare the predicted values obtained from EPIWIN with PBT profiler results. By doing this, the reproducibility of EPIWIN predictions can be tested since the PBT estimates are based on EPIWIN.

As expected, the available QSAR-predictions from the EPIWIN suite are in line with the one obtained from PBT profiler (Log  $K_{ow}$ , Henry's Law Constant and Environmental Partitioning (MacKay (III), see Table 6).

Required Property	Predicted Value using PBT profiler	Predicted Value using EPIWIN																													
Water solubility	420 mg/l	486 mg/l at 25 °C																													
Partition coefficient n-octanol/water	log $K_{ow}$ = 1.4	log $K_{ow}$ = 1.4																													
Bioconcentration Factor (BCF)	BCF = 71	BCF = 71																													
Henry's Law Constant	0.000000029 atm-m <sup>3</sup> /mole	2.86 *10 <sup>-9</sup> atm-m <sup>3</sup> /mole																													
Photolysis (Atmospheric OH Rate Constant)	Rate constant =0.00000000031	Rate Constant = 31. 46 E-12 cm <sup>3</sup> /molecule-sec																													
Ultimate biodegradation	2.75 ( weeks)	2.75 (weeks-months)																													
<b>Environmental Partitioning (MacKay, (III))</b>	<table border="1"> <thead> <tr> <th></th> <th>Mass Amount (percent)</th> <th>Half-Life (days)</th> </tr> </thead> <tbody> <tr> <td>Air</td> <td>1</td> <td>0.5</td> </tr> <tr> <td>Water</td> <td>35</td> <td>15</td> </tr> <tr> <td>Soil</td> <td>64</td> <td>30</td> </tr> <tr> <td>Sediment</td> <td>0</td> <td>140</td> </tr> </tbody> </table>		Mass Amount (percent)	Half-Life (days)	Air	1	0.5	Water	35	15	Soil	64	30	Sediment	0	140	<table border="1"> <thead> <tr> <th colspan="2">Level III Fugacity Model:</th> </tr> <tr> <th>Mass Amount (percent)</th> <th>Half-Life (hr)</th> </tr> </thead> <tbody> <tr> <td>Air</td> <td>1</td> </tr> <tr> <td>Water</td> <td>35</td> </tr> <tr> <td>Soil</td> <td>64</td> </tr> <tr> <td>Sediment</td> <td>0</td> </tr> <tr> <td></td> <td>3.24*10<sup>3</sup></td> </tr> </tbody> </table>	Level III Fugacity Model:		Mass Amount (percent)	Half-Life (hr)	Air	1	Water	35	Soil	64	Sediment	0		3.24*10 <sup>3</sup>
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	3.24*10 <sup>3</sup>																														

**Table6:** Comparison between the results obtained from EPIWIN with PBT profiler.

### Step 5: Assess your substance to identify whether it is PBT or vPvB

The assessment of persistence is based on the degradation data (biotic and chemical) and on the half life data.

Based on available data for structurally-similar compounds AES, Sodium laureth sulphate is expected to biodegrade rapidly in the environment. This is in contrast to the predictable results by BIOWIN (see table 5). However the assessment of biodegradability will be based on the experimental data since it is available.

Moreover, the predicted rate constant and estimated half-life for the reaction of hydroxyl radicals with Sodium laureth sulphate in the atmosphere indicated that when Sodium laureth sulphate is released to the atmosphere is likely to be degraded (not persistence) by this fate process. As a result, these data suggest that Sodium laureth sulphate is unlikely to be persistent. In addition, both PBT Profiler and EPIWIN v3.2 estimate for Level III Fugacity Model have estimated that Sodium laureth sulphate is expected to be found predominantly in soil. However, its half-life in soil is 30 days, does not exceed the EU criteria of >120 days. Therefore, Sodium laureth sulphate is estimated to be not persistent in the environment.

As a conclusion, based on the available data (measures and predicted) degradation and half life, Sodium laureth sulphate is not considered to meet the screening criteria for persistence.

#### **Is your substance bioaccumulative?**

No measured BCF data is available. The potential for bioaccumulation was therefore assessed on the basis of a calculated BCF.

Both PBT profiler and BCFWIN v2.17 predict BCF of 71 (<2000, not accumulative) based on logK<sub>OW</sub> of 1.4.

Based on this value Sodium laureth sulphate is not considered to meet the screening criteria for bioaccumulation.

#### **Is your substance toxic to the environment organisms?**

Aquatic toxicity data for structurally similar compounds have been identified in several sources (HERA risk assessment, Masden et al, 2001). The acute toxicity values L(E)C50 for fish and Daphnia reported for AES ranged between 0.39 mg/L to 450 mg/L and 1 and 50 mg/L respectively. These values are clearly above 0.1 mg/l and therefore Sodium laureth sulphate is not considered to meet the screening criteria for aquatic toxicity. In addition, Ying claimed that most surfactants are not acutely toxic to organisms at environmental concentrations and their aquatic chronic toxicity occurs usually at concentrations above than 0.1mg/l apart of alkylphenols surfactants.

In conclusion, based on the available aquatic toxicity data, Sodium laureth sulphate is not considered fulfilling the T screening criterion related to aquatic toxicity.

**Is your chemical classified as potential PBT?**

Based on the available data, the substance has been shown to have a half-life of 30 days <180 days in soil and biodegrades fast. Therefore the substance does not fulfil the P-criterion. The estimated BCF for the substance in fish is 71 (<2000) and therefore the substance does not fulfil the B-criterion. The reported aquatic toxicity data for Sodium laureth sulphate suggests that the substance does not fulfil the T-criterion related to aquatic toxicity.

Therefore on the basis of the predictions and measured data the **substance should not be considered as potential PBT.**

**What is the preferred environmental compartment of your chemical?  
(Environmental Distribution )**

**Adsorption:**

The substance has a high  $K_{oc}$  value of 3498 indicating that the substance is likely to adsorb onto soil. This is also in agreement with the Mackay level III predictions, which shows that that soil is the preferred environmental compartment (64%). Therefore the persistence of Sodium laureth sulphate in soil is probably more significant than the persistence in water.



